Remarkable Carbonylation of a Co-ordinated Phospha-alkyne Bu^tCP to afford the Novel Phosphinidene Ligand Bu^tC(CO)P, Crystal and Molecular Structure of $[Re_2(CO)_8Pt(dppe){Bu^tC(CO)P}]$ (dppe = 1,2-bisdiphenylphosphinoethane)

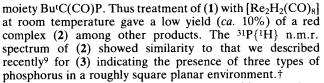
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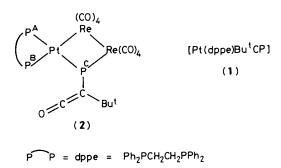
An unusual carbonylation of a co-ordinated phospha-alkyne to give a phosphinidine is reported.

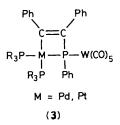
In a series of papers¹⁻⁸ we have described the variety of bonding modes exhibited by the phospha-alkyne ligands $RC\equiv P$ ($R = Bu^t$, adamantyl) in transition metal complexes. Previously we showed that the η^2 -bonding mode of the co-ordinated phospha-alkyne in [Pt(dppe)Bu^tCP] (1) changed on treatment with [Fe₂(CO)₉] to a μ^3 , $\eta^2 \perp$ type in the complex [Pt(dppe)(Bu^tCP)Fe₂(CO)₆].²

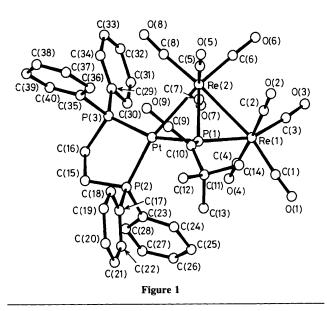
We now report an unusual reaction in which the coordinated Bu^tCP in (1) is attacked by carbon monoxide to produce a complex containing the remarkable phosphinidine



A single crystal structure determination on (2) revealed the unusual molecular structure shown in Figure 1, in which the







† ³¹P N.m.r. data for (**2**): δ P^A – 84.1, P^B –93.0, P^C + 142.0 p.p.m.; J(PtP^A) 2710; J(PtP^B) 3590, J(PtP^C) 2120, J (P^AP^B) 0, J(P^AP^c) 267, J(P^BP^C) 23 Hz.

Bu^tC(CO)P generated by CO attack at the C atom of Bu^tCP acts as a bridge between platinum and one of the rhenium atoms. \ddagger

As expected the P(1)-C(10) bond length [1.83(2) Å in (2)] is typical of a single bond and the P(1)C(10)C(11), C(9)C(10)C(11), and P(1)C(10)C(9) bond angles are close to 120° .

Although structures (2) and (3) appear to be related using the isolobal relationship $\text{Re}(\text{CO})_4 \leftrightarrow \mathbb{O}$ CR suggesting the presence of an Re=Re double bond in (2), the observed Re(1)-Re(2) distance [3.044(1) Å] is almost identical with the single bond length observed in $[\text{Re}_2(\text{CO})_{10}]^{10}$ of 3.041(1) Å. Furthermore it should be noted that whereas the environment around phosphorus in (2) is planar, in (3) it is approximately tetrahedral.⁹ A further interesting feature of the structure of (2) is the relatively short transannular P(1)-Re(2) distance [2.371(6) Å] compared with P(1)-Re(1) [2.453(6) Å].

The attack of carbon monoxide on a co-ordinated phosphaalkyne to generate the R(CO)P fragment is of interest since it may be the first step in the coupling of two phospha-alkynes

‡ Crystal Data: C₄₀H₃₃O₉P₃PtRe₂, M = 1318.1, monoclinic, space group P2₁/c, a = 14.356(4), b = 12.012(2), c = 25.570(6) Å, $\beta = 94.01(2)^\circ$, U = 4398.7 Å³, Z = 4, $D_c = 1.99$ g cm⁻³. The structure was solved by heavy atom methods and refined to R = 0.060 using 3812 reflections with $I > \sigma(I)$ measured on an Enraf-Nonius CAD4 diffractometer. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1. RC=P (R = adamantyl), via carbon monoxide which we observed recently⁷ in the formation of the ligand PCR(CO)PCR and further studies of this type of reaction are planned.

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References

- 1 J. C. T. R. Burckett-St. Laurent, P. B. Hitchcock, H. W. Kroto, and J. F. Nixon, J. Chem. Soc., Chem. Commun., 1981, 1141.
- 2 S. I. Al-Resayes, P. B. Hitchcock, M. F. Meidine, and J. F. Nixon, J. Chem. Soc., Chem. Commun., 1984, 1080.
- 3 R. Bartsch, J. F. Nixon, and N. Sarjudeen, J. Organomet. Chem., 1985, 294, 267.
- 4 M. F. Meidine, C. J. Meir, S. Morton, and J. F. Nixon, J. Organomet. Chem., 1985, 297, 255.
- 5 M. F. Meidine, J. F. Nixon, and R. Mathieu, J. Organomet. Chem., 1986, 314, 307.
- 6 P. B. Hitchcok, M. J. Maah, and J. F. Nixon, J. Chem. Soc., Chem. Commun., 1986, 737.
- 7 P. B. Hitchcock, M. J. Maah, and J. F. Nixon, J. Chem. Soc., Chem. Commun., 1987, 658.
- 8 R. Bartsch, P. B. Hitchcock, M. F. Meidine, and J. F. Nixon, J. Organomet. Chem., 1984, 266, C41.
- 9 D. Carmichael, P. B. Hitchcock, J. F. Nixon, F. Mathey, and A. Pidcock, J. Chem. Soc., Chem. Commun., 1986, 762.
- 10 M. R. Churchill, K. N. Amoh, and H. J. Wasserman, *Inorg. Chem.*, 1981, 20, 1609.